## 2N Sodium Hydroxide Solution, Method 1

## I. Required Items:

10N Sodium Hydroxide
Deionized water $\left(\mathrm{diH}_{2} \mathrm{O}\right)$

$$
\begin{array}{cc}
\text { For } 100 \mathrm{~mL} \\
20 \mathrm{~mL} & \frac{\text { For } 500 \mathrm{~mL}}{80 \mathrm{~mL}}
\end{array}
$$

## II. Instructions For Preparation:

Combine 10 N sodium hydroxide with $\mathrm{diH}_{2} \mathrm{O}$. The volume prepared should be only the necessary volume for appropriate waste disposal.

## III. Storage Conditions:

N/A - Only enough solution is made for immediate use.

## IV. Expiration Date:

N/A - Only enough solution is made for immediate use.

| Vendor/Lot\# <br> 10N Sodium <br> Hydroxide | Vender/Lot\# <br> diH2 $_{2} \mathrm{O}$ | Amount <br> Made | Date <br> Prepared | Preparer | Expiration <br> Date |
| :--- | :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |

