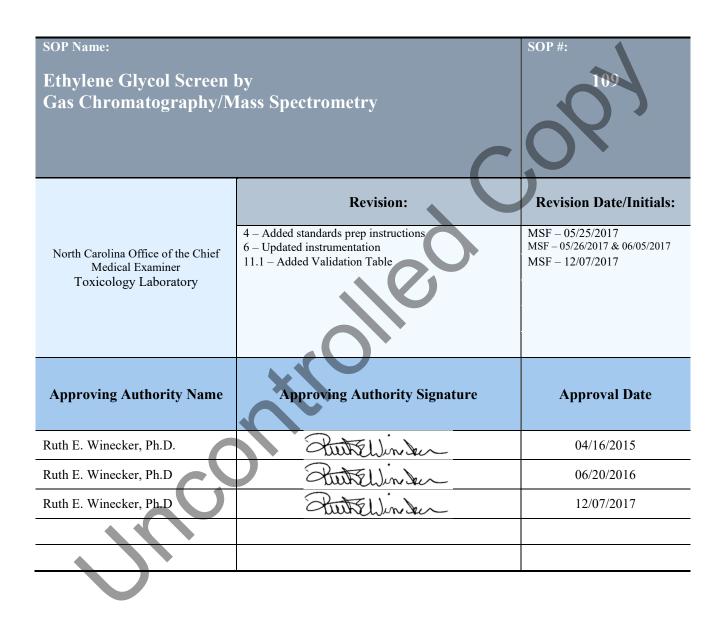
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## 1. Principle of Assay

- 1.1. Ethylene glycol is a common component of commercial anti-freeze preparations, automotive cooling systems, and hydraulic brake fluids. This, coupled with its sweet taste, results in its relatively common appearance in emergency departments and medical examiner laboratories.
- 1.2. Ethylene glycol is completely miscible with water, and as such is extremely difficult to remove from blood and serum samples. This procedure describes a method which extracts the ethylene glycol present in blood or serum, such that it can be easily derivatized and identified by gas chromatography/mass spectrometry.

## 2. Specimens

2.1. This procedure is applicable to serum, plasma, whole blood, vitreous humor, and urine specimens. A 100µL sample size is generally employed unless a dilution is required so that the calibration curve encompasses the expected range of the unknown specimens.

## 3. Reagents and Materials (HPLC grade)

- 3.1. Ethylene Glycol
- 3.2. 1,2-Butanediol
- 3.3. Acetone
- 3.4. Deionized water
- 3.5. Methylene Chloride
- 3.6. Phenylboronic acid
- 3.7. 2,2-dimethoxypropane

### 4. Standards, Controls, and Solutions

4.1. Prepare Calibration, IS, and Control standards, as needed, according to SOP 10.

### 4.2. 1,2-Butanediol Internal Standard (100 mg/dL)

- 4.2.1. Into a 10mL volumetric flask, add 0.05mL of 1,2-Butanediol with a micropipette.
- 4.2.2. Fill to the line with DiH20, insert stopper and invert three times to mix. Transfer to properly labeled 16x100mm screw topped test tubes and cap. Store in laboratory refrigerator (R1-2601). See SOP-010.

#### 4.3. Ethylene Glycol Stock Calibration Standard (10000mg/dL)

- 4.3.1. Place a 10mL volumetric flask containing 1-2mL of DiH<sub>2</sub>0 onto a rough balance and tare.
- 4.3.2. Using a pasture pipette, transfer 1g Ethylene Glycol, dropwise, into the flask.
- 4.3.3. Fill to the line with DiH<sub>2</sub>0, insert stopper and invert three times to mix.
- 4.3.4. Transfer to properly labeled 16x100mm screw topped test tubes and cap. Store in laboratory refrigerator (R1-2601). See <u>SOP-010</u>.

### 4.4. Ethylene Glycol Calibration Standard (Cal 2-6 - 100mg/dL)

- 4.4.1. Into a 10mL volumetric flask, add 0.1 ml of Ethylene Glycol Stock Calibration Standard (10000mg/dL) with a class: A volumetric pipette.
- 4.4.2. Fill to the line with  $DiH_20$ , insert stopper and invert three times to mix. Transfer to properly labeled 16x100mm screw topped test tubes and cap. Store in laboratory refrigerator (R1-2601). See <u>SOP-010</u>.

### 4.5. Ethylene Glycol Stock Control Standard (10000mg/dL)

- 4.5.1. Place a 10mL volumetric flask containing 1-2mL of DiH<sub>2</sub>0 onto a rough balance and tare.
- 4.5.2. Using a pasture pipette, transfer 1g Ethylene Glycol, dropwise, into the flask.
- 4.5.3. Fill to the line with DiH<sub>2</sub>0, insert stopper and invert three times to mix.
- 4.5.4. Transfer to properly labeled 16x100mm screw topped test tubes and cap. Store in laboratory refrigerator (R1-2601). See <u>SOP-010</u>.

### 4.6. Ethylene Glycol Control Standard (100mg/dL)

- 4.6.1. Into a 10mL volumetric flask, add 0.1 ml of Ethylene Glycol Stock Control Standard (10000mg/dL) with a class: A volumetric pipette.
- 4.6.2. Fill to the line with DiH<sub>2</sub>0, insert stopper and invert three times to mix. Transfer to properly labeled 16x100mm screw topped test tubes and cap. Store in laboratory refrigerator (R1-2601). See <u>SOP-010</u>.

### 4.7. 1% Phenylboronic acid solution

4.7.1. Weigh 1g Phenylboronic acid, transfer to 100 mL volumetric flask, and qs to volume with acetone. Store at 2-8°C for up to 1 year.

### 4.8. 0.1% Phenylboronic acid solution

- 4.8.1. Pipette 10 mL of 1% Phenylboronic acid solution into a 100 mL volumetric flask and QS to volume with acetone. Make fresh daily.
- 4.9. Phenylboronic Acid (5mg/mL) in 2,2-dimethoxypropane
  - 4.9.1. To a 100 mL volumetric flask, add 0.500 grams of phenylboronic acid, and fill to the mark with 2,2-dimethoxypropane. PROTECT FROM MOISTURE.

## 5. Equipment and Special Supplies

- 5.1. Test tubes, 16x100 borosilicate glass
- 5.2. Micropipette  $(10-100\mu L)$
- 5.3. Pipette tips  $-(10-100\mu L)$
- 5.4. Vortex mixer
- 5.5. Centrifuge (2000-3000 rpm)
- 5.6. GC autosampler vials
- 5.7. Nitrogen Evaporator
- 5.8. Pasteur Pipettes

## 6. Instrumentation

- 6.1. Agilent GC/MS 6890/5973, Chemstation software, compatible computer and printer.
- 6.2. GCMS13 Instrument Method: "13ETHYLENEGLYCOL.M"
  - 6.2.1. Click here for instrument parameters
- 6.3. GC/MS parameters (The instrument and conditions may be changed to permit improved performance).
  - 6.3.1. Acquisition mode: SIM
  - 6.3.2. Column: DB5-MS
  - 6.3.3. MS transfer line temperature: 280° C
  - 6.3.4. Oven ramp program:

6.3.4.1.	Equilibration time:	30 sec.
6.3.4.2.	Initial temp:	70° C
6.3.4.3.	Initial temp hold:	1.00 min.
6.3.4.4.	Ramp 1:	25° C/min.
6.3.4.5.	Ramp 1 final temp:	175° C
6.3.4.6.	Ramp 1 hold:	0.00 min.
6.3.4.7.	Ramp 2:	35° C/min.
6.3.4.8.	Ramp 2 final temp:	310° C
6.3.4.9.	Ramp 2 hold:	0.5 min.
6.3.4.10.	Total Run Time:	9.56 min.
6.3.5. Inlet Parameters		COX
6.3.5.1.	Mode:	SPLIT
6.3.5.2.	Split ratio:	30:1
6.3.5.3.	Split vent:	51.0mL/min

- 6.3.5.4. Temperature:
- 6.3.5.5. Injection Volume:
- 6.3.5.6. Purge time:
- 6.3.5.7. Column Flow: 1.7mL/min (Constant Flow)

Constant

## 7. Target Ions

7.1. Major Ions for Ethylene glycol phenyl borate ester: 91, 148, 118

7.2. Major Ions for 1,2-Butanediol phenyl borate ester: 147, 91

## 8. Procedure

- 8.1. Label disposable 16x100 mm test tubes for each standard, blank, control and specimen.
- 8.2. Add 50μL of the 1,2-butanediol internal standard solution to all tubes.
- 8.3. Add the ethylene glycol standard solution and ethylene glycol quality control to appropriately labeled tubes.
- 8.4. Add 100  $\mu$ L of negative blood bank blood to all standards, controls and blanks.

- 8.5. Add 100  $\mu$ L of unknown specimen to appropriately labeled tubes.
- 8.6. Add 300  $\mu$ L of acetone to each tube to deprote inate blood, vortex, and let sit for 3-5 minutes.
- 8.7. Centrifuge for 10 minutes at 2000-3000 x g.
- 8.8. Transfer 100 μL of supernatant to a clean, appropriately labeled 5 mL conical test tube.
- 8.9. Add 100 μL of 5μg/mL phenylboronic acid in 2,2-dimethoxypropane, and vortex.
- 8.10. Centrifuge for 5 minutes at 2000-3000 RPM.
- 8.11. Transfer 50 μL into a clean, labeled autosampler tube, and inject on the GC/MS.(Note: if two layers are formed in step 10, transfer only the top layer.)

## 9. Calculations

- 9.1. Calibration
  - 9.1.1. A one point calibration curve is used.
- 9.2. Dilution Factor

9.2.1. D = Sample volume (curve) / Sample volume (unknown)

9.3. Multiplier for homogenates/dilutions and non-standard volumes

9.3.1.  $M = (V_{curve} / V_{samp}) \times D$ 

- 9.3.1.1. M = Multiplier
- 9.3.1.2. D = dilution factor
- 9.3.1.3.  $V_{curve}$  = matrix volume of calibration curve
- 9.3.1.4.  $V_{samp} = matrix volume of specimen$
- 9.4. Concentration

9.4.1. C = (A / V) \* M

9.4.1.1. C = Concentration (ng/mL) of the analyte in the unknown case.

9.4.1.2. A = Amount of drug in sample
9.4.1.3. V = Volume of sample
9.4.1.4. M = Multiplier

### 9.5. Qualifier Ion Ratios

- 9.5.1. Ratio  $1 = QL_1/QN$ 
  - 9.5.1.1.  $QL_1$  = response of the quantifying product ion
  - 9.5.1.2. QN = response of the qualifying product ion

## 10. Quality Control

- 10.1. For an analysis to be acceptable the following criteria must be met:
  - 10.1.1. Chromatography must be acceptable with a symmetrical (Gaussian) shape. Each analyte of interest must have near baseline resolution from any other peaks in the chromatogram.
  - 10.1.2. The retention time of each analyte should be within  $\pm 5\%$  of the expected retention time based on the calibrators and the relative retention time to the internal standard.
  - 10.1.3. The quality control samples shall have an analytical value as stated on the standard and control worksheet, and shall not deviate  $\pm 30\%$  from the expected value.
  - 10.1.4. The blank shall represent a specimen of "none detected" and should not contain analyte signal with appropriate ion ratios above 10% of the low standard.
  - 10.1.5. The internal standard areas of the samples shall not deviate more than 50-200% from the average internal standard areas of the calibrators.
  - 10.1.6. Calculated ion ratios shall not deviate more than 20% to that of the average ion ratios of the calibrators.
  - 10.1.7. Results are reported qualitatively.

### 11. Validation of Method

Parameter	Result
Bias (all matrix - blood, urine, hd liver)	L: -10.86% H: -9.37%
Precision (all matrix - blood, urine, hd liver)	L: 2.21% H: 0.83%
Calibration model	Linear: Equal weighting
Carryover	No carryover observed following high calibrator.
Interference Studies	Not evaluated -
Ion Ratios	lon ratios were noted to be concentration dependent. Ion ratio range for unknowns shall be determined using the closest calibrator concentration ion ratio(s) as the target.
Ionization/Suppression: (Not needed if IS coelutes within 0.05 min.)	N/A - IS coelutes with target analyte (0.1 min).
LOD (Determined experimentally)	1.35 mg/dL
LOQ (Set to lowest calibrator with acceptable Bias/Precision).	10 mg/dL
Dilution Integrity	Average % difference of diluted specimens is 3.51% - lowest: 0.44% Highest 6.63% (n=8)
Processed Sample Stability - (re-analyze after 8 days)	Not Determined
Recovery	Not Determined
11.1.	

## 12. Load Assignment Packet Preparation

- 12.1. After completing all data generation and reviewing for corrections, the analyst will assimilate the data in the following order:
  - 12.1.1. Load assignment sheets, followed by any additional notes to file pertaining to load.

- 12.1.2. Load specimen sheet.
- 12.1.3. Data summary
- 12.1.4. Chain of Custody.
- 12.1.5. Standard Sheet.
- 12.1.6. GC/MS Method and Calibration Report(s).
- 12.1.7. Running Sequences.
- 12.1.8. Chromatograms and results for calibrators, controls and specimens
- 12.1.9. The Load Checklist should be initialed and dated to acknowledge completion of load. The finished data package will be placed in the data review box.

### 13. References

13.1.Baselt, Randall C. *Disposition of Toxic Drugs and Chemicals in Man*. Foster City, CA: Chemical Toxicology Institute, 2000. 406-08. Print.